

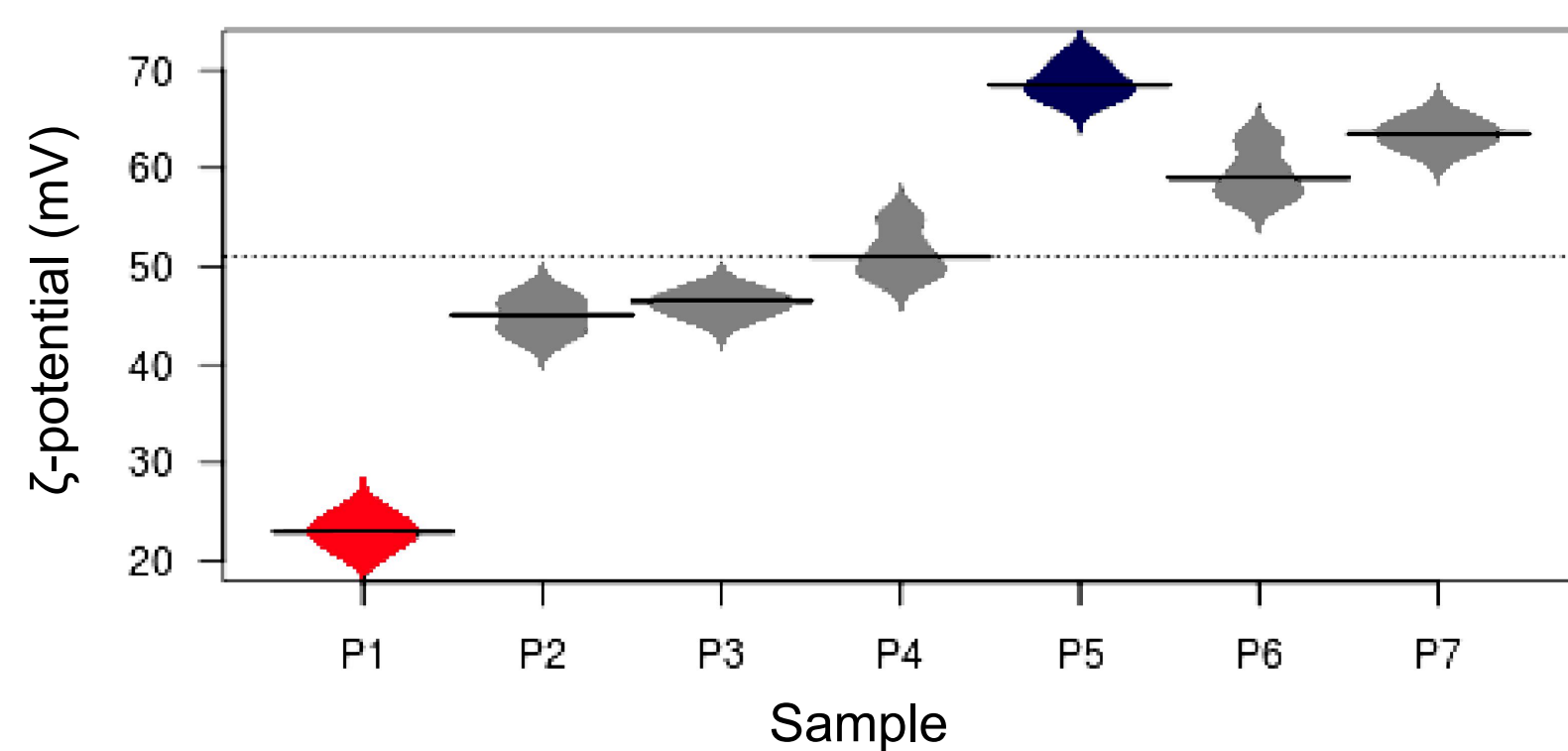
## INTRODUCTION

- The MAPLE deposition technique can produce uniform films, composed of species with different molecular weights, with unique morphologies using the advantages of vacuum deposition.
- The purpose of this work was to obtain new multifunctional coatings based on chitosan, bioglass, ZnO and graphene oxide.
- In this study, the deposition was carried out on 73Ti-20Zr-5Ta-2Ag.
- The suspensions used for deposition were studied in detail, determining the stability of the particles in suspension.
- The best combination of deposition parameters will be presented.

## SAMPLE COMPOSITIONS

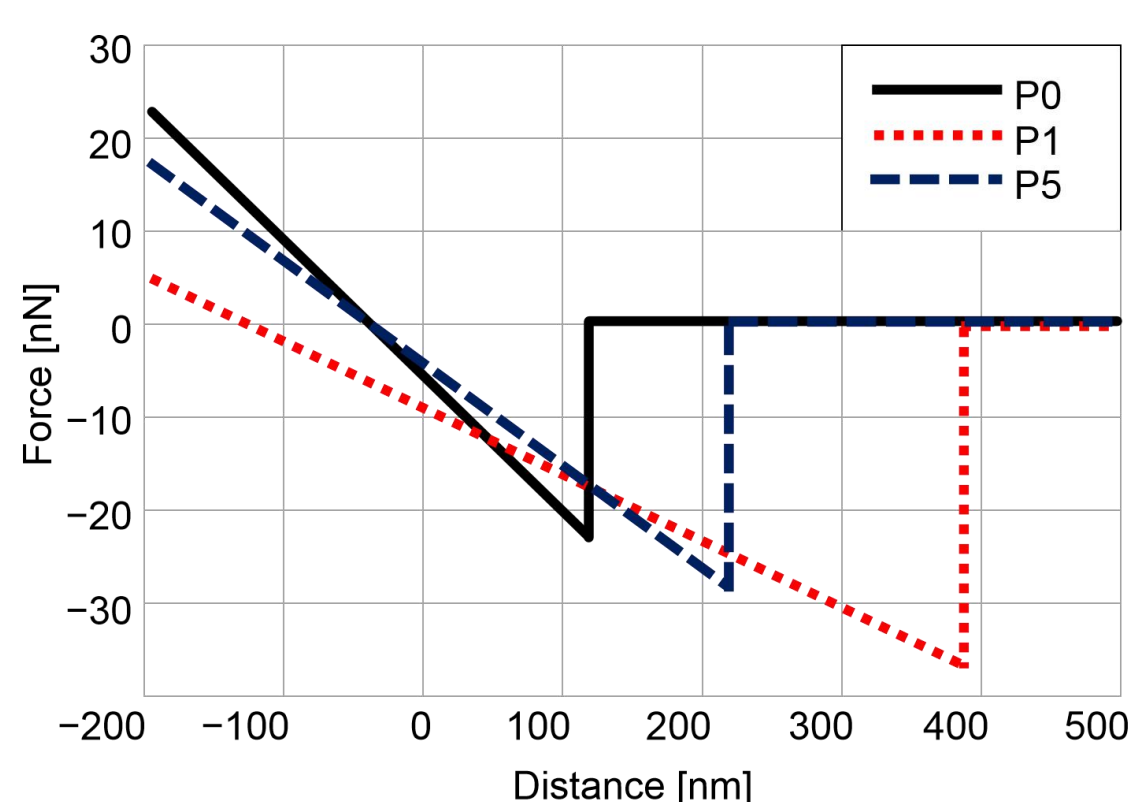
Sample	Components	Chi:(BG:ZnO:GO) ratio (wt./wt./wt.)	BG:ZnO:GO ratio (wt./wt./wt.)
P0	Ti-Zr-Ta-Ag	-	-
P1	Ti-Zr-Ta-Ag/Chi	-	-
P2	Ti-Zr-Ta-Ag/Chi/(BG/ZnO/GO)	1:1	5:1:1
P3	Ti-Zr-Ta-Ag/Chi/(BG/ZnO/GO)	1:1	1:1:1
P4	Ti-Zr-Ta-Ag/Chi/(BG/ZnO/GO)	1:1	1:5:1
P5	Ti-Zr-Ta-Ag/Chi/(BG/ZnO/GO)	2:1	1:5:1
P6	Ti-Zr-Ta-Ag/Chi/(BG/ZnO/GO)	2:1	1:1:1
P7	Ti-Zr-Ta-Ag/Chi/(BG/ZnO/GO)	2:1	5:1:1

## ZETA POTENTIAL



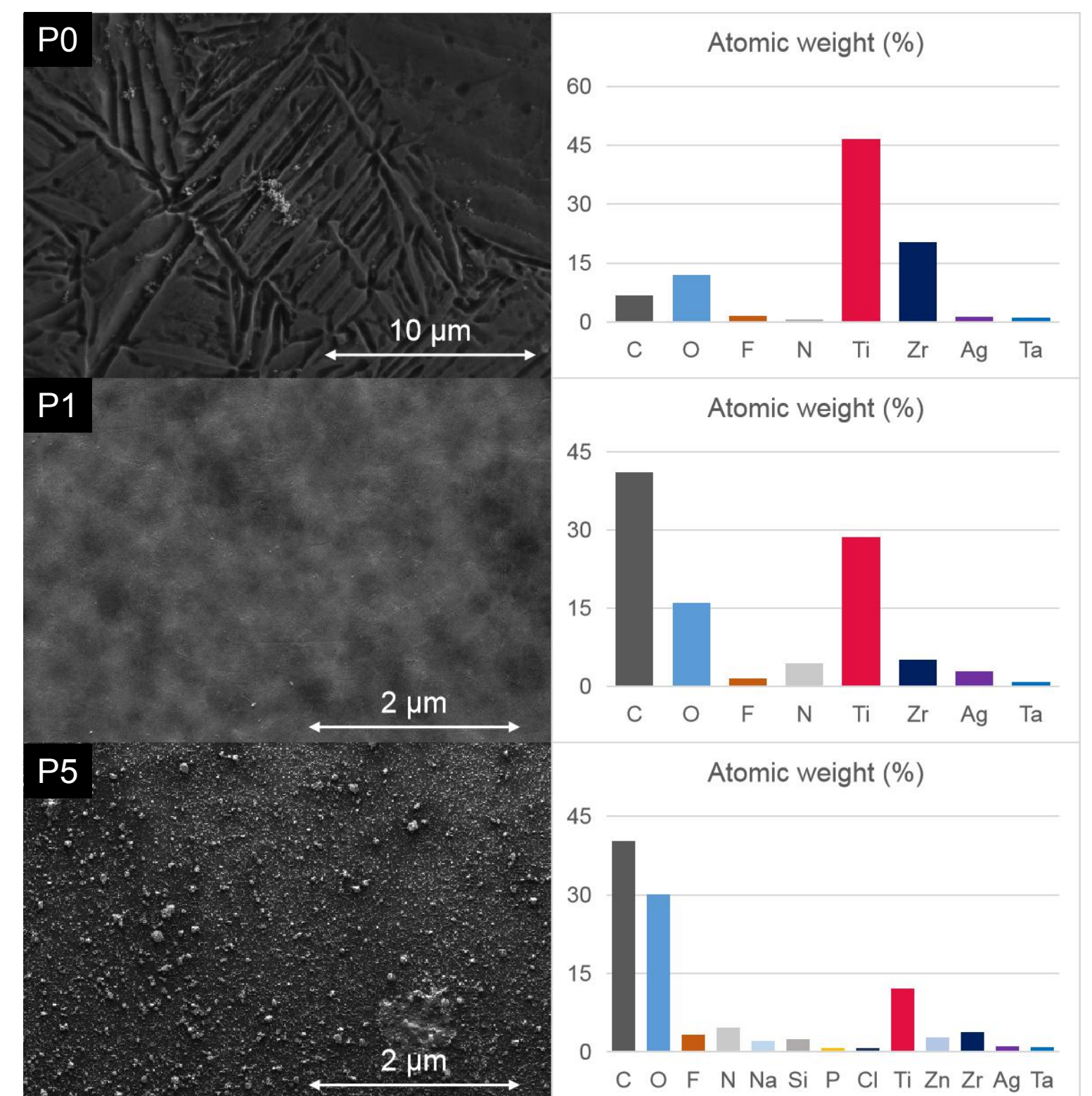
- Particles with ζ-potentials higher than +30 mV or lower than -30 mV are considered to have moderate stability, while high stability is generally ensured by ζ-potential values >50 mV.
- These values can ensure a sufficient electrostatic repulsion to generate stability over a longer period of time.
- All the mixtures had a good stability, however a slight decrease in stability was observed with the increase of the ceramic component, this behavior being due to the large dimensions of the BG.
- The addition of ZnO led to an increase in stability.

## SURFACE ADHESION FORCES



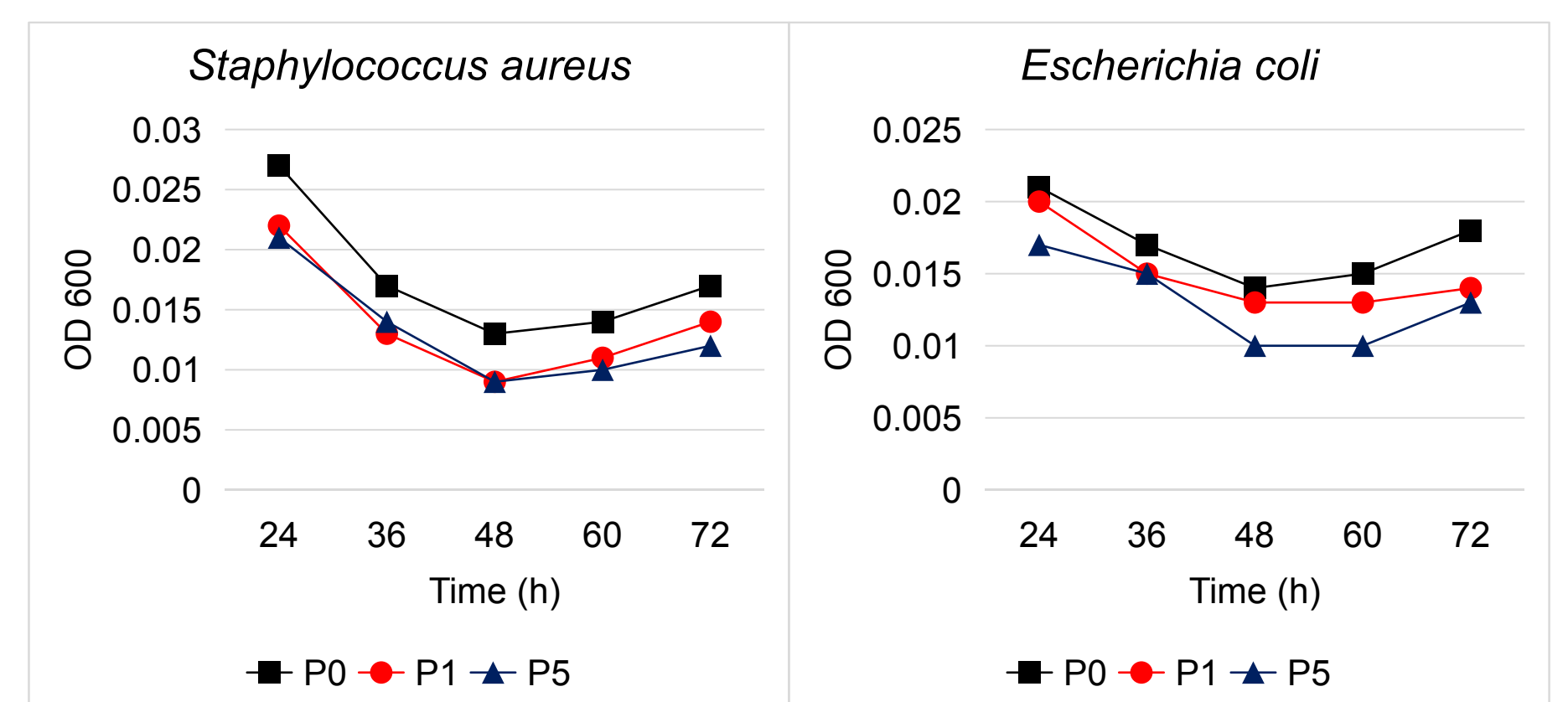
- Sample P0 had the highest adhesion force of  $24 \pm 3$  nm and a short activity range, which can be attributed to the relatively smooth and tough oxide surface.
- Sample P1 had the lowest adhesion force of  $6 \pm 2$  nm, but it had the longest range of activity, probably given by the semi-elastic properties of the chitosan film deposited on the surface.
- Sample P5 gave mixed results depending on the material that the AFM tip touched during the experiments, with a median value of  $16 \pm 4$  nm being between the values of samples P1 and P2.

## SURFACE MORPHOLOGY



- Sample P0 surface is covered with complex oxides resulting from the etching process.  $0.7-1 \mu\text{m}$  silver formations are present, scattered on the surface.
- Sample P1 had a surface uniformly covered with the chitosan film. No oxide formations were found to protrude from the film.
- On Sample P5, the ceramic particles are well dispersed in the polymeric matrix, with sizes between  $0.1$  and  $0.5 \mu\text{m}$ .
- The EDX spectra revealed that the surface of sample P0 is mainly composed of metallic oxides produced after the etching process.
- Sample P1 shows the increase in carbon percentage from the coating with the chitosan solution.
- On sample P5 all the components from the mixture are present on the surface.

## ANTIBACTERIAL ACTIVITY



- The first reading of OD600 was made after 24 h, followed by subsequent readings at the specified time intervals.
- It was observed that the OD600 starts to decrease continuously for all samples in the interval of 24-48 h, and then it begins to slowly rise again for the next 24 h.
- The most effective antibacterial activity for all materials was observed after 48 h of incubation.
- As the OD600 is lower, the bacterial growth is also low.
- The slow rise in OD600 after 60 and 72 h means that the materials' antibacterial activity starts to decrease as time passes, the maximum activity being at 48 h.

## ACKNOWLEDGEMENT

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